

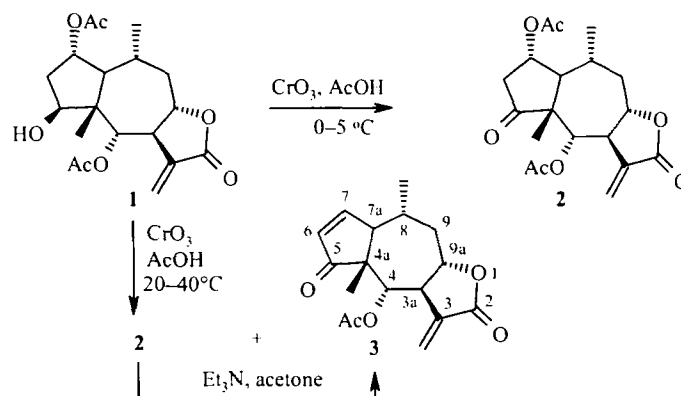
## LETTERS TO THE EDITOR

### OXIDATION OF BRITANIN

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**Keywords:** sesquiterpene lactones, britanin, bigelovin, oxidation.

The oxidation products of the sesquiterpene lactone britanin (**1**) from *Inula britannica* L. are of considerable interest as starting compounds for obtaining a whole range of physiologically active substances [1]. Also literature data on its oxidation are ambiguous. Under identical conditions, dehydrobritanin (**2**) [2] or bigelovin (**3**) [3] have been obtained by treating lactone **1** with chromium oxides. We have investigated this reaction and established that formation of compounds **2** and **3** is controlled by the reaction temperature. Oxidation of lactone **1** by  $\text{CrO}_3$  in acetic acid at a reaction temperature of 0–5°C quantitatively leads to the dehydro derivative **2**; if the temperature is raised to 20–40°C, mixtures of compounds **2** and **3** are always formed. The data obtained were confirmed by HPLC analysis (column 4 × 250 mm, packed with 5 μm Diasorb 130 C16/T, the eluent was an acetonitrile gradient in water; detection by UV at 254 nm) and  $^1\text{H}$  NMR spectra of the isolated compounds and reaction mixtures. Vigorous stirring of equimolar amounts of britanin and  $\text{CrO}_3$  in AcOH at 20°C for 3 h leads to a 1:1 mixture of compounds **2** and **3**, and the content of the latter increases slightly when the reaction time is increased. Lactone **3** is formed quantitatively only in dehydroacetylation of compound **2** by treatment with bases.



**4-Acetoxy-3,3a,4,4a,7a,8,9,9a-octahydro-4a,8-dimethyl-3-methyleneazuleno[6,5-b]furan-2,5-dione (Bigelovin, 3).**  $\text{CrO}_3$  (54.6 mg, 0.55 mmol) was added in small portions to a solution of lactone **1** (200 mg, 0.55 mmol) in acetic acid (3 ml). The mixture was vigorously stirred at 20°C for 3 h, diluted with water (15 ml), and extracted with chloroform (5 × 10 ml). The extract was dried over  $\text{Na}_2\text{SO}_4$  and evaporated under vacuum. The residue was boiled for 2 h in acetone (10 ml) containing triethylamine (0.42 g, 4.1 mmol). The oxidation product was purified on silica gel; eluent hexane–acetone, 2:1. Obtained 157 mg (94%) of bigelovin; mp 191–192°C;

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according to data in [4], mp 190-191°C. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 1.2 (3H, s, 4a-CH<sub>3</sub>); 1.3 (3H, d, *J* = 7.2 Hz, 8-CH<sub>3</sub>); 1.54 (1H, q, *J* = 11.8 Hz, 9-H); 1.95 (3H, s, CH<sub>3</sub>CO); 2.03 (1H, m, 9-H); 2.58 (1H, m, 8-H); 3.05 (2H, m, 3a-H and 7a-H); 4.6 (1H, ddd, *J*<sub>1</sub> = 11.6, *J*<sub>2</sub> = 10.5, *J*<sub>3</sub> = 3.0 Hz, 9a-H); 5.58 (1H, d, *J* = 7.5 Hz, 4-H); 5.9 (1H, d, *J* = 3.1 Hz, CH<sub>2</sub>=); 6.1 (1H, dd, *J*<sub>1</sub> = 6.0, *J*<sub>2</sub> = 2.9 Hz, 6-H); 6.2 (1H, d, *J* = 3.5 Hz, CH<sub>2</sub>=); 7.7 (1H, dd, *J*<sub>1</sub> = 6.0, *J*<sub>2</sub> = 1.8 Hz, 7-H). Found, %: C 66.64; H 6.22. C<sub>17</sub>H<sub>20</sub>O<sub>5</sub>. Calculated, %: C 67.11; H 6.58.

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